

RESEARCH PAPER

## Structural and Optical Properties of Graphene/Chlorophyll Pigment Nanocomposite

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### ABSTRACT

Compared with their bulky or nanosized counterparts, nanostructured carbon materials attract many researchers due to their diverse practical applications. Graphene has been successfully fabricated using a pulsed laser ablation process. It relies on the exfoliation of a graphite disc to induce polycarbonate sheets suspended in an aqueous medium (deionized water). Chlorophyll was extracted from spinach and then mixed with graphene. The properties were studied using a scanning electron microscope (SEM) and X-ray diffraction (XRD). Various graphene nanostructures have been observed, such as disks, dots, and fibers. The optical properties were studied by UV-vis spectroscopy and photoluminescence (PL) measurements. PL of graphene typically shows strong photoemission in the visible region (400.7 nm, 675.3 nm).

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### INTRODUCTION

Graphene is known as an atomic layer of graphite, which is also the essential unit for fullerenes and CNTs. It is a two dimensional (2D) crystal that is stable under ambient conditions [1, 2]. Graphene has exceptional in-plane structural, mechanical, thermal and electrical properties. These properties make it attractive for application in many research fields [3, 4]. Optoelectronic devices based on graphene have garnered significant interest and show potential for use in solar cells, touch screens, and photodetectors [5-10]. The exceptional optical characteristics of graphene, such as linear optical absorption [11, 12], tunable band-gap [13], and intrinsic photocurrent have been established [14-17]. These qualities can be combined with other unique mechanical and electrical characteristics of graphene to

provide novel functions. The weakness of pristine graphene's absorption, which can reach 2.3%, severely restricts its use in photodetection.

As a result, other strategies have been used to improve the interaction between graphene and photons, such as plasmonic effects [18, 19], photothermoelectric effects [20], sensitization by quantum dots [10, 21, 22], or bulk semiconductors [23]. In this work, we employ chlorophyll as a light-absorbing material for graphene phototransistors and examine the hybrid graphene-chlorophyll devices' optoelectronic performance. As an organic semiconductor, chlorophyll absorbs light very well. Furthermore, chlorophyll is an exceptionally stable and abundant biomaterial [24]. Studies on chlorophyll photosensitization could be useful for organic photovoltaics [25, 26], optical sensors [27], and artificial photosynthesis [28], since chlorophyll

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is essential to the process of photosynthesis. Furthermore, because chlorophyll-related molecules come in a variety of forms, it is possible to tailor the energy level alignment in hybrid graphene-organic molecule systems to maximize charge separation and transfer. There are two types of chlorophyll—types A and B—found in green algae and terrestrial plants. The presence of methyl in chlorophyll a, which is substituted by a formyl group in chlorophyll b, is what distinguishes these two types of chlorophylls. In higher plants, the ratio of chlorophyll a to chlorophyll b is roughly 3:1. The visible spectrum's red (650–700 nm) and

blue–violet (400–500 nm) bands are the primary wavelengths that chlorophyll absorbs [29].

## MATERIALS AND METHODS

### Pulse laser ablation

Graphene powders can be obtained by using Q-Switched Nd: YAG pulsed laser in liquids as showed in Fig. 1.

### Preparation of chlorophyll (collection of plants and chlorophyll extraction)

The spinach leaves from the plant is collected. Ten milliliters of 80% acetone were used to grind

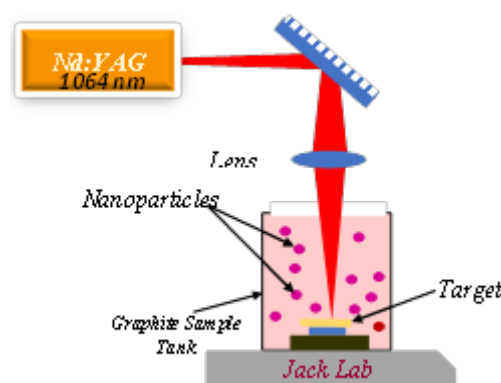


Fig. 1. Schematic diagram of PLAL method to produce graphene.

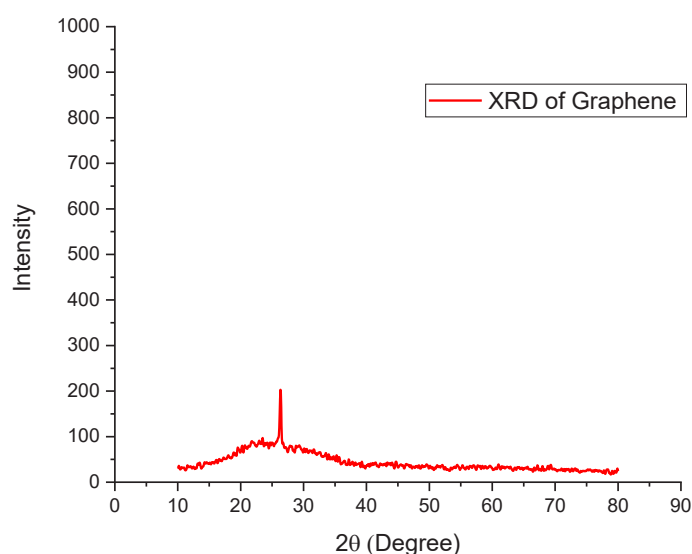


Fig. 2. XRD pattern of graphene.

five grams of freshly chopped leaves. Then, for five minutes, it was centrifuged at 5000–10000 rpm. Once the supernatant was moved, the process was repeated until the residue lost all of its color [29].

## RESULTS AND DISCUSSION

### XRD pattern

A significant peak at  $2\theta = 26^\circ$  can be seen in the

XRD pattern of graphene, as illustrated in Fig. 2. It's a unique peak that appear to be quite broad with lesser intensity when compared to bulk graphite's patterns, which is consistent with the previous report [30].

### Morphological study

Using FESEM, the surface morphology was examined. The graphene nanoparticles have a

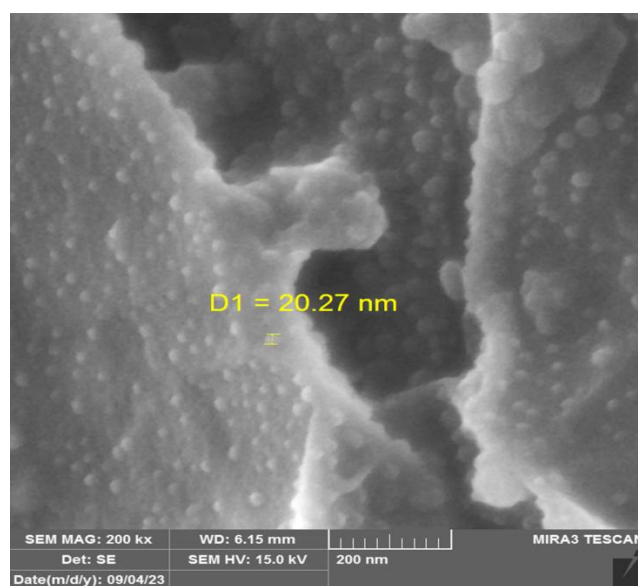


Fig. 3. FE-SEM image of graphene powders.

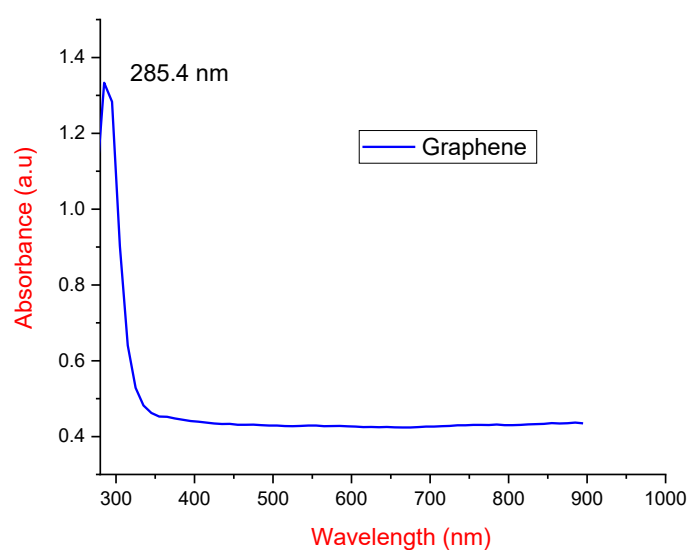


Fig. 4. UV-Vis spectrum of graphene.

spherical shape and different sizes and have an almost homogeneous distribution as showed in Fig. 3. The Image J program was used to calculate the average particle size and it was about 40 nm, which is close to what the researcher reported in the reference [31].

#### Optical properties

##### UV-visible spectrum

The UV-visible spectrum of graphene is

displayed in Fig. 4. According to earlier research, graphene has an absorption peak at 285.4nm [32]. The electrical arrangement of graphene during the reduction of graphene oxide is the cause of this peak. The  $n-\pi^*$  transition of C–O bonds, which are now embedded on the graphene due to exfoliation and intercalation, is responsible for this absorption peak.

Fig. 5 illustrates the UV-Vis spectrum of chlorophyll extract dissolved in acetone

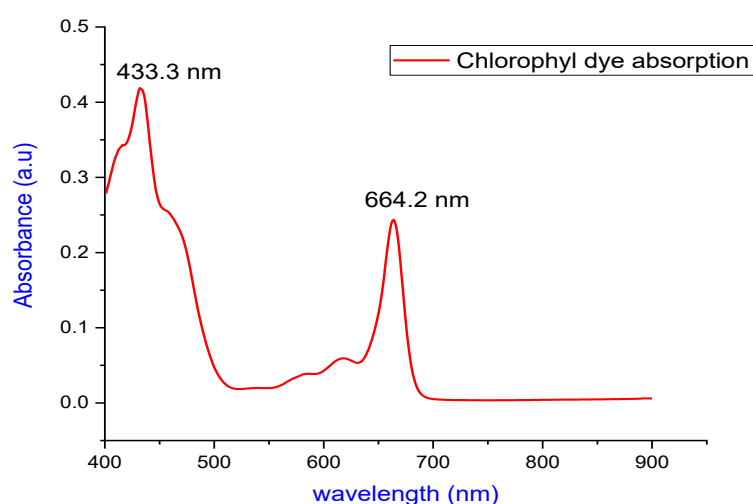


Fig. 5. UV-Vis spectrum of chlorophyll.

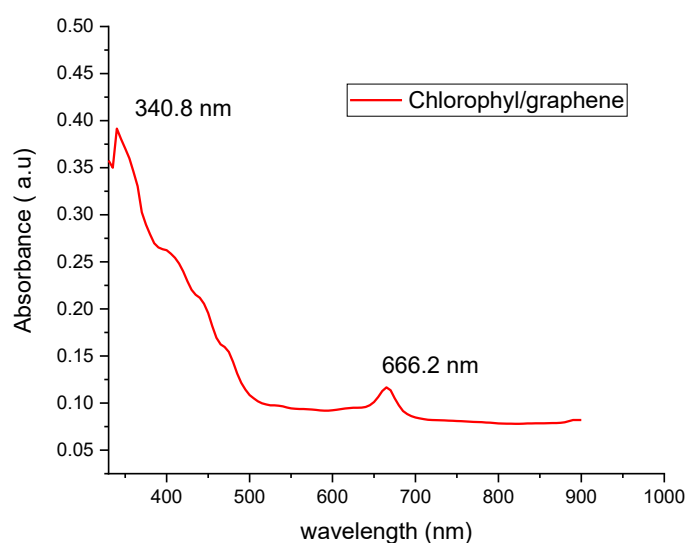


Fig. 6. UV-Vis spectrum of graphene/chlorophyll sample.

absorption. The highest absorption was seen at 433.3 and 664.2 nm [33].

Fig. 6 shows the UV-Vis spectrum of graphene/

chlorophyll sample. The absorption peaks for the chlorophyll/graphene are located at 340.8 and 666.2 nm.

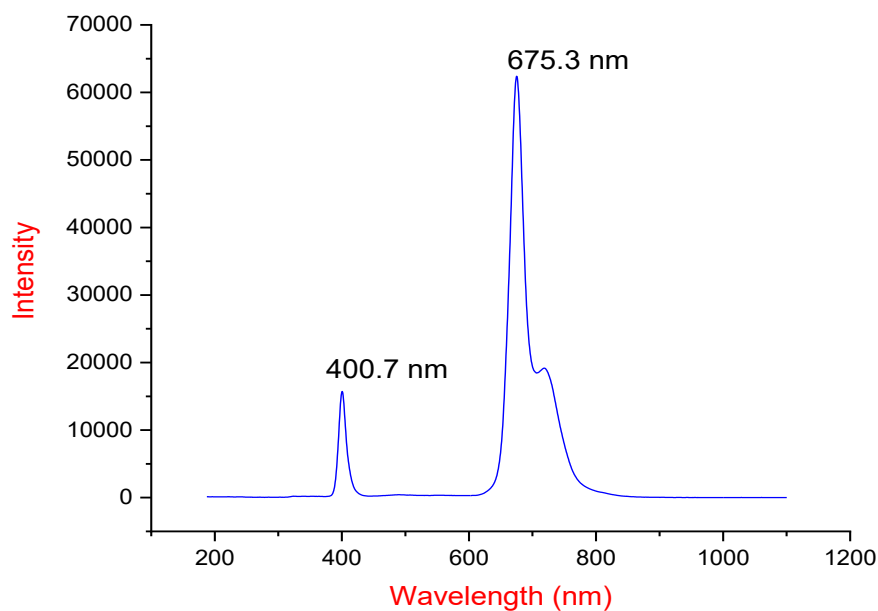


Fig. 7. PL spectrum of graphene nanoparticles.

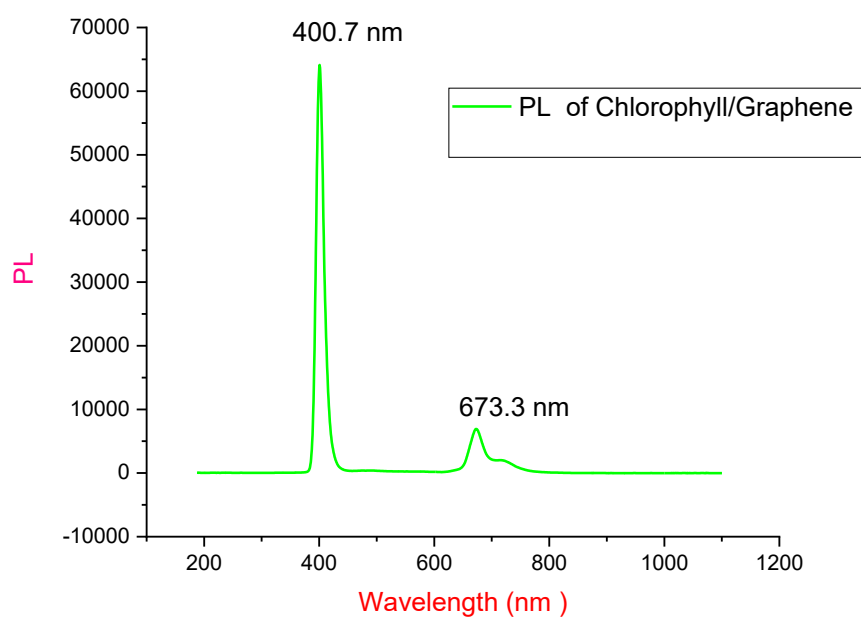


Fig. 8. PL spectrum of graphene/chlorophyll sample.

### Photoluminescence

The researchers [34] affirmed that the emission peak of carbon is usually wide with large Stokes shift as compared with that of organic dyes and it may result from the wide distributions of differently sized particles and surface chemistry, different emissive traps (salvation effect), or a mechanism currently unresolved [35]. The emission of the graphene typically shows wide optical emission in the visible region (Fig. 7).

Also, the additional PL data of graphene/chlorophyll sample is shown in Fig. 8. In this sample, there is a clear quenching of PL. The PL quenching suggests that charge transfer occurs between the chlorophyll film and graphene, which reduces radiative recombination since electron-hole pairs are produced in the chlorophyll molecules under illumination [36].

### CONCLUSION

The pulsed laser, with a wavelength of 1064 nm, energy of 80 mJ, a frequency of 6 Hz, and a number of pulses of 400 pulses, has the ability to generate nanoparticles having a spherical shape and homogeneous distribution with absorption peak at 285.4 nm. The pulsed laser has the ability to exfoliate graphite layers through the appearance of a single peak of low intensity as compared to graphite. Through the PL results, we notice that mixing chlorophyll with graphene nanoparticles enhanced the emission spectrum at 400.7 nm, and this indicates the removal of a small portion of impurities during the mixing process and showed quenching of graphene as a result of charge transfer from chlorophyll, which caused a reduction in radiative recombination.

### CONFLICT OF INTEREST

The authors declare that there is no conflict of interests regarding the publication of this manuscript.

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