## **RESEARCH PAPER**

# Green Synthesize of Zinc Oxide Nanoparticles and Nano-Fluids with Pomegranate Extract Used in Mouthwash Solutions

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# ABSTRACT

Zinc oxide nanoparticles were synthesized with a green procedure applying pomegranate extract. Hydrothermal reactions with the help of stainless steel autoclave at various temperatures and times were used. Results confirmed by adjusting temperature and time reaction, uniform star-like and flowerlike zinc oxide nanostructures were synthesized. For preparation of mouthwash nano-fluid, nanoparticles with the aid of ultra-sound wave (200W, 60 min) irradiation were dispersed to the pomegranate extract. The phase of prepared products were examined by X-ray diffraction pattern (XRD), band gap and optical were measured by UV-visible absorption spectroscopy, the bonds using the (FTIR) spectrometry and morphology via scanning electron microscopy (SEM). The antibacterial exam of flower like nano-structures was performed applying the non-growth halo (diffusion disk) test. Streptococcus mutans bacteria were chosen as suitable bacteria for this investigation.

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#### **INTRODUCTION**

The use of nanoparticles in various household applications and various industries has dramatic changes in reducing costs and increasing efficiency. In nanotechnology, with decreasing dimensions, the surface-to-volume ratio increases greatly, which reduces the consumption of materials. And have critical sizes and allowable limits have been considered[1-5].

Also, as we know, with smaller dimensions than the critical size, new properties and properties \* Corresponding Author Email: Msnr88827@yahoo.com are observed for nanoparticles that can be shown by manipulating atoms to show new properties and properties. Due to the increasing problems of pollution in wastewater, the use of green and biocompatible materials instead of toxic chemicals has received much attention in recent decades [6-10].

New syntheses have recently been reported using plant and fruit extracts. The use of extracts is biocompatible, cost-effective and innovative. Traditionally produced mouthwashes use a

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variety of chemicals if many of the pathogens and potential carcinogens can be eliminated instead of using hazardous and harmful chemicals made from green methods. Zinc oxide is a semiconductor, non-toxic substance used in many electronics and cosmetics. The aim of this study was to make zinc oxide nanoparticles using pomegranate extract and nanofluid in aqueous solution and used as a mouthwash. is a round bacterium, facultatively anaerobic, gram-positive coccus, it is a main contributor to tooth decay usually found in the human oral cavity [11-18].

## MATERIALS AND METHODS

Materials and Physical Measurements

All of the chemicals were used as received



Fig. 1. Schematic of green synthesis of zinc oxide with hydrothermal reaction



Fig. 2. XRD pattern of ZnO nanostructures

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without further purifications. All precursors were purchased from Merck and Sigma- Aldrich. A multiwave ultrasonic generator (Sonicator 3000; Bandeline, MS 72, Germany), equipped with a converter/transducer and titanium oscillator (horn), 12.5 mm in diameter, operating at 20 kHz



Fig. 3. SEM images of ZnO nano-stars at (a,b) 120°C (c,d) 180°C (e,f) 200°C

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with a maximum power output of 100 W, was used for the ultrasonic irradiation. The ultrasonic generator automatically adjusted the power level. The wave amplitude in each experiment was adjusted as needed. XRD patterns were recorded by a Philips, X-ray diffractometer using Ni-filtered Cu K $\alpha$  radiation. SEM images were obtained using a LEO instrument model 1455VP. Prior to taking images, the samples were coated with a very thin layer of Pt to make the sample surface conducting and prevent charge accumulation, and obtaining a better contrast. FT-IR spectra were recorded on Galaxy series FTIR5000 spectrophotometer. Room temperature photoluminescence was studied by a Perkin Elmer fluorescence instrument.

#### Green synthesize of zinc oxide nanoparticles

Firstly 1g of zinc nitrate was dissolved in 100 ml of pomegranate extract with mixing on the stirrer. Then the solution was put to autoclave reactor for different times from 2 to 24h at 100 to 200 OC. Finally the white precipitate was washed and centrifuged and was dried at oven fOr 24 h at 60 OC. Fig. 1 shows green synthesis of zinc oxide with hydrothermal reaction schematically.

For preparation of mouthwash nano-fluid, nanoparticles with the aid of ultra-sound wave (200W, 60 min) irradiation were dispersed to the pomegranate extract.

### **RESULTS AND DISCUSSION**

Fig. 2 illustrate XRD pattern of ZnO nanostructures, as a results confirmed the pattern has suitable agreement with standards peaks (JCPDS code: 79-0208). It is indexed as a pure hexagonal structure with suitable agreement to literature value (Space group: P63mc).

Figs. 3a, 3b show scanning electron microscopy images of ZnO nano-structures at 120°C that confirmed formation of nano-flower structures. Figs 3c, 3d illustrate product that were synthesized at 180°C that approve star-like structure were obtained. Figs 3e, 3f depict nano-products that were made at 200°C which results also confirm nano-star zinc oxide were achieved.

Fig. 4 illustrates FT-IR spectrum of zinc oxide nano structures with pomegranate extract, as we expected there are some weak absorption because of presence of organic compound (pomegranate extract) on the zinc oxide nanoparticles.



Fig. 4. FT-IR spectrum of zinc oxide nano structures with pomegranate extract

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Fig. 5. Disc diffusion antibacterial test against Streptococcus mutans(a) ZnO prepared at  $120^{\circ}$ C (b) ZnO prepared at  $180^{\circ}$ C , c) ZnO prepared at  $200^{\circ}$ C

The antibacterial test of nanomaterials was performed using the non-growth halo test (diffusion disk). Streptococcus mutans bacteria were chosen as suitable bacteria for this test. As the Fig 5 confirm, we see the lack of growth of bacteria around the disc containing nanoparticles.

#### CONCLUSION

ZnO nanostructures were prepared by a simple hydrothermal process at low temperature. The effect of different surfactants such as time and temperature on the morphology of zinc oxide nanostructures was investigated. The antibacterial test of nanomaterials was performed using the non-growth halo test (diffusion disk). Streptococcus mutans bacteria were chosen as suitable bacteria for this test. Nanostructures were characterized by X-ray diffraction, scanning electron microscopy. Fourier transform infrared spectrometer, the purity of the material was also determined.

#### **CONFLICT OF INTEREST**

The authors declare that there is no conflict of interests regarding the publication of this manuscript.

#### REFERENCES

- Hassanpour M, Safardoust H, Ghanbari D, Salavati-Niasari M. Microwave synthesis of CuO/NiO magnetic nanocomposites and its application in photo-degradation of methyl orange. Journal of Materials Science: Materials in Electronics. 2015;27(3):2718-2727.
- 2. Saffarzadeh S, Nabiyouni G, Ghanbari D. Preparation of  $Ni(OH)_{2^{\prime}}$  NiO and  $NiFe_2O_4$  nanoparticles: magnetic and photo-catalyst  $NiFe_2OV$  –NiO nanocomposites. Journal of Materials Science: Materials in Electronics. 2016;27(12):13338-13350.
- 3. Wu X, Cai J, Li S, Zheng F, Lai Z, Zhu L, et al. Au@Cu<sub>2</sub>O stellated polytope with core–shelled nanostructure for high-performance adsorption and visible-light-driven photodegradation of cationic and anionic dyes. Journal of Colloid and Interface Science. 2016;469:138-146.
- Masoumi S, Nabiyouni G, Ghanbari D. Photo-degradation of azo dyes: photo catalyst and magnetic investigation of CuFe<sub>2</sub>O<sub>4</sub>-TiO<sub>2</sub> nanoparticles and nanocomposites.

Journal of Materials Science: Materials in Electronics. 2016;27(9):9962-9975.

- Buthiyappan A, Abdul Aziz AR, Wan Daud WMA. Recent advances and prospects of catalytic advanced oxidation process in treating textile effluents. Rev Chem Eng. 2016;32(1):1-47.
- Aliyan H, Fazaeli R, Jalilian R. Fe<sub>3</sub>O<sub>4</sub>@mesoporous SBA-15: A magnetically recoverable catalyst for photodegradation of malachite green. Appl Surf Sci. 2013;276:147-153.
- Liu S, Sun H, Ang HM, Tade MO, Wang S. Integrated oxygendoping and dye sensitization of graphitic carbon nitride for enhanced visible light photodegradation. Journal of Colloid and Interface Science. 2016;476:193-199.
- Bakre PV, Volvoikar PS, Vernekar AA, Tilve SG. Influence of acid chain length on the properties of TiO<sub>2</sub> prepared by sol-gel method and LC-MS studies of methylene blue photodegradation. Journal of Colloid and Interface Science. 2016;474:58-67.
- Alhaji MH, Sanaullah K, Lim S-F, Khan A, Hipolito CN, Abdullah MO, et al. Photocatalytic treatment technology for palm oil mill effluent (POME) – A review. Process Saf Environ Prot. 2016;102:673-686.
- Ghanbari D, Salavati-Niasari M, Ghasemi-Kooch M. In situ and ex situ synthesis of poly(vinyl alcohol)–Fe<sub>3</sub>O<sub>4</sub> nanocomposite flame retardants. Particuology. 2016;26:87-94.
- 11. Modern Ferrite Technology. Springer US; 2006.
- 12. Kinemuchi Y, Ishizaka K, Suematsu H, Jiang W, Yatsui

K. Magnetic properties of nanosize  $NiFe_2O_4$  particles synthesized by pulsed wire discharge. Thin Solid Films. 2002;407(1-2):109-113.

- 13. Fesharaki MJ, Ghanbari D. Photo-catalyst Fe–Pt nanocomposite: mechanical preparation of iron nanoparticles and simple synthesis of platinum nanoparticles. Journal of Materials Science: Materials in Electronics. 2017;28(13):9804-9812.
- 14. Ghanbari D, Salavati-Niasari M. Synthesis of urchin-like CdS-Fe<sub>3</sub>O<sub>4</sub> nanocomposite and its application in flame retardancy of magnetic cellulose acetate. Journal of Industrial and Engineering Chemistry. 2015;24:284-292.
- Chandradass J, Jadhav AH, Kim KH, Kim H. Influence of processing methodology on the structural and magnetic behavior of MgFe<sub>2</sub>O<sub>4</sub> nanopowders. J Alloys Compd. 2012;517:164-169.
- O'Donnell LA. Sherris Medical Microbiology as a Resource for Doctor of Pharmacy StudentsReview of: Sherris Medical Microbiology, 5thedition; Kenneth Ryan et al.; (2010). McGraw Hill, New York, NY. 1026 pages. Journal of Microbiology and Biology Education. 2012;13(1):102-103.
- Elliott T. Medical microbiology: Edited by S. BARON. 1986, 2nd ed. Addison-Wesley Publishers Ltd, Wokingham, Berks. Pp. xxvi and 1262. 19.95. J Med Microbiol. 1986;22(3):284-284.
- Thomas VJ, Rose FD. Ethnic differences in the experience of pain. Social Science & amp; Medicine. 1991;32(9):1063-1066.